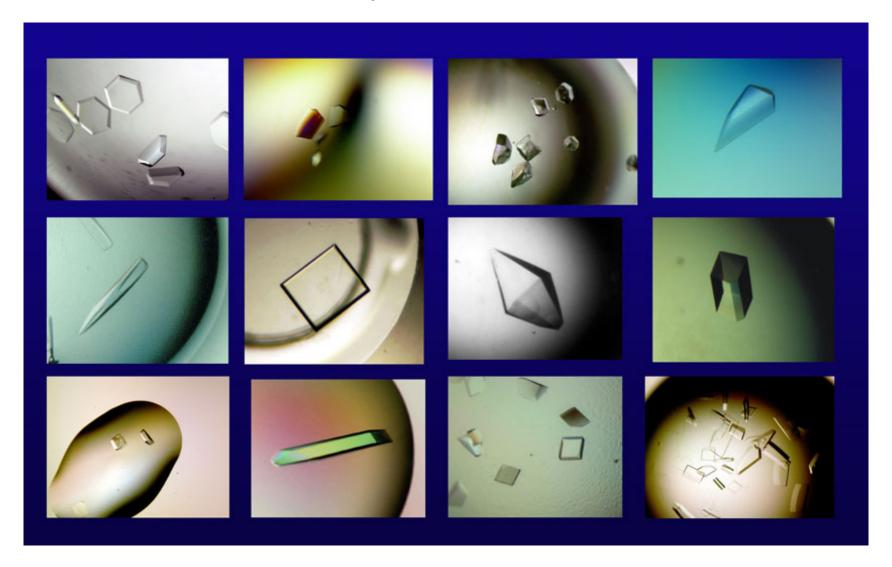
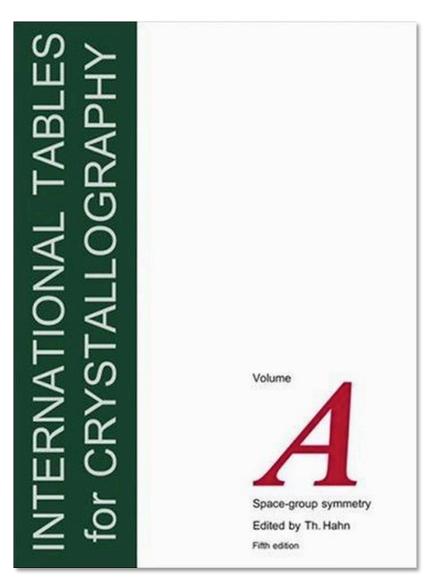
# Introduction to symmetry

Andrey Lebedev, CCP4

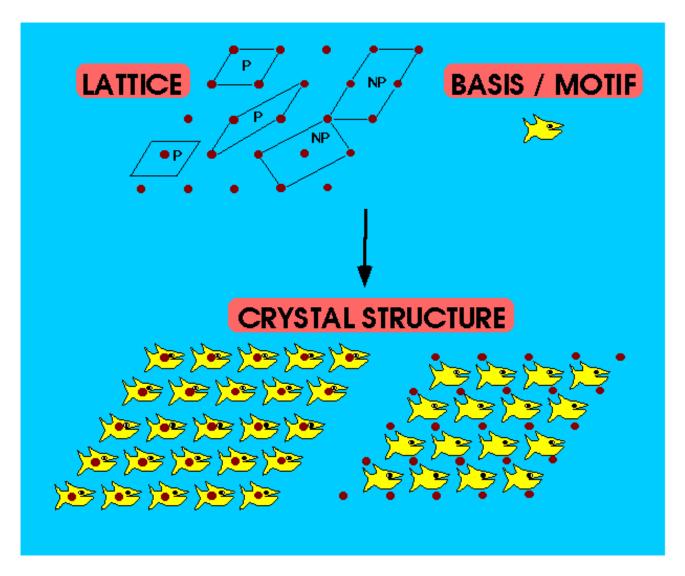


#### The Reference

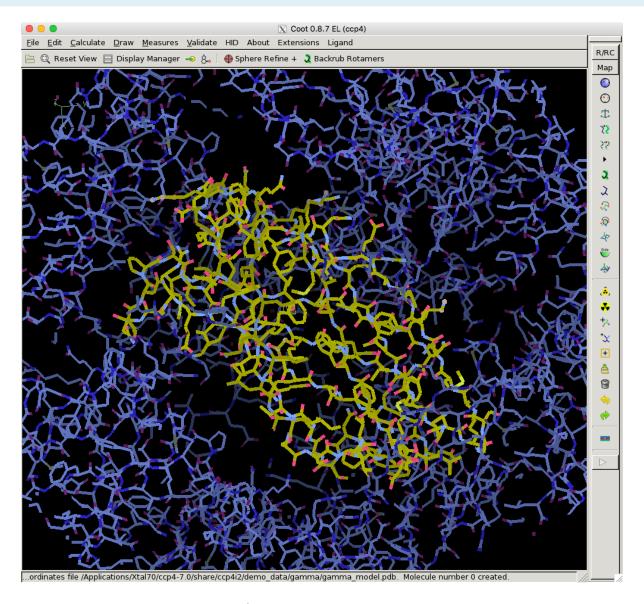


## Crystal: unit cell + lattice

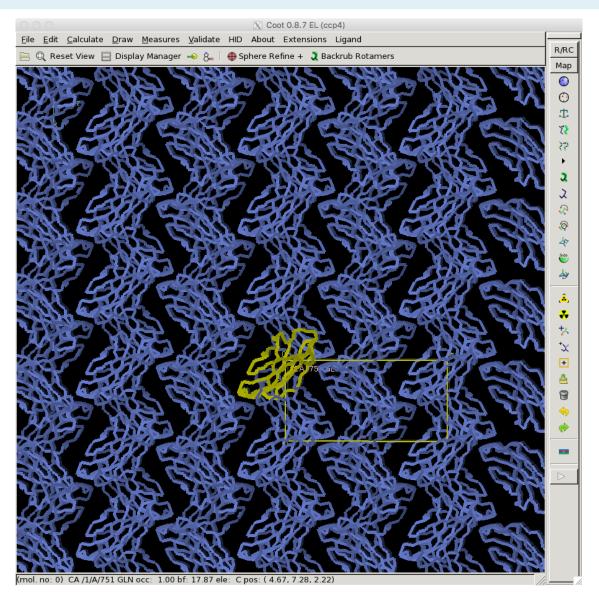
Conventional (constructive) definition of crystal structure.

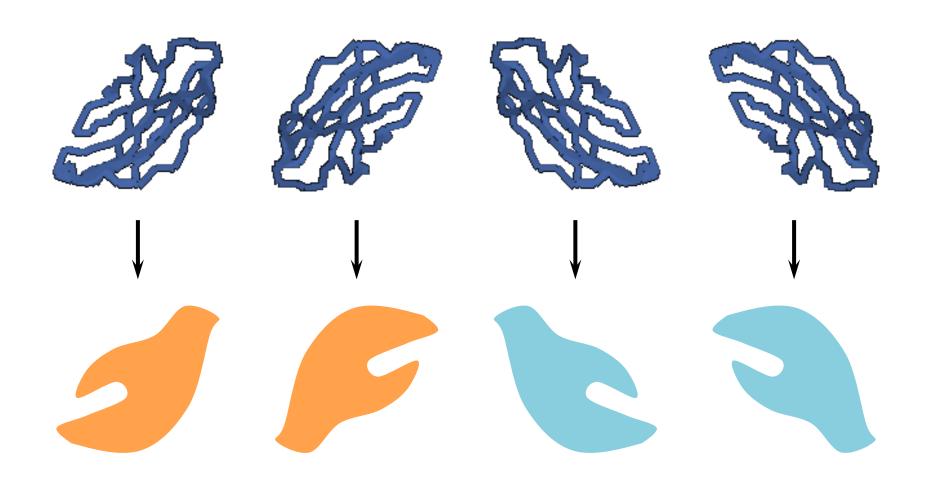


### Good for start? Examine symmetry using Coot

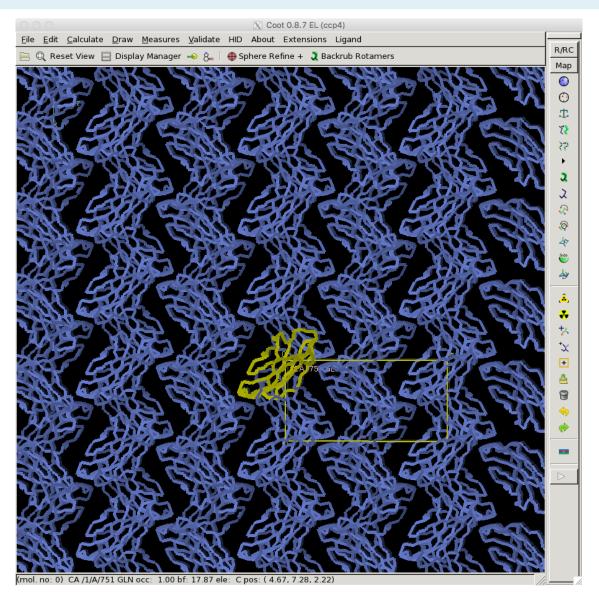


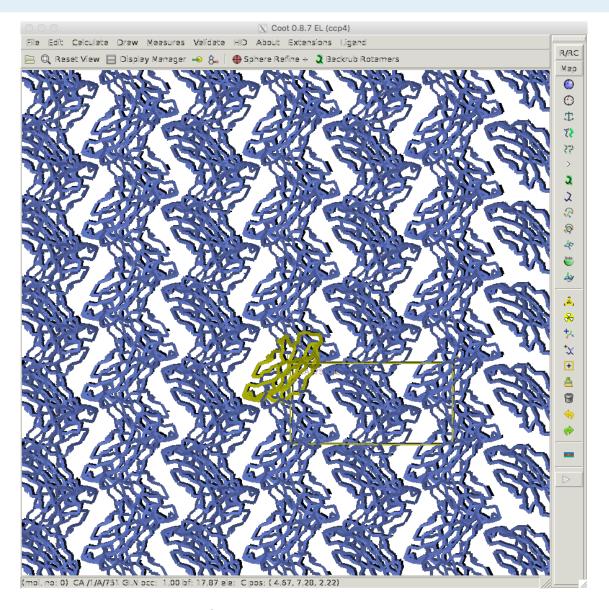
## Symmetry view in Coot

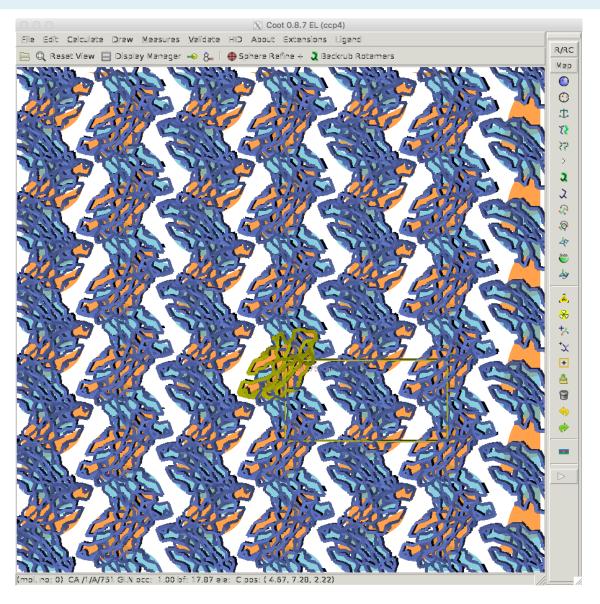


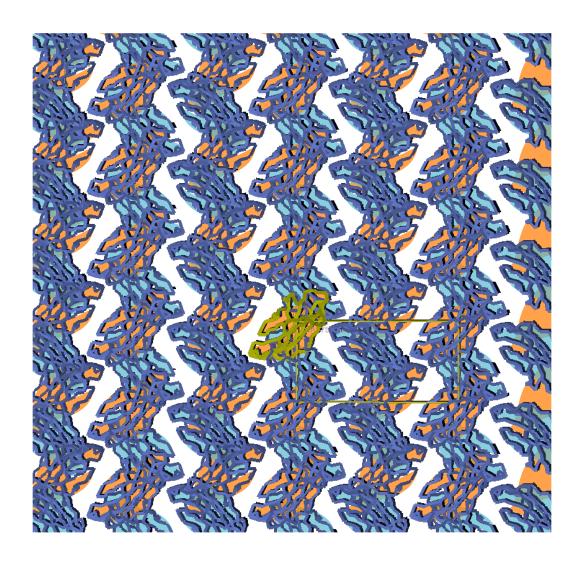


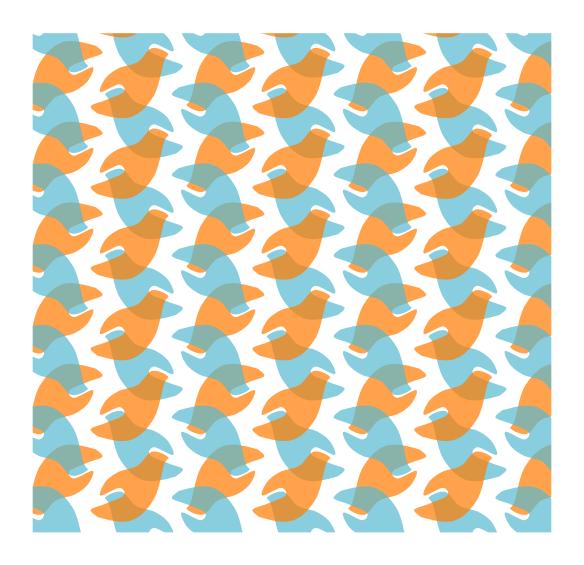
Opposite sides of molecules a denoted with different colours

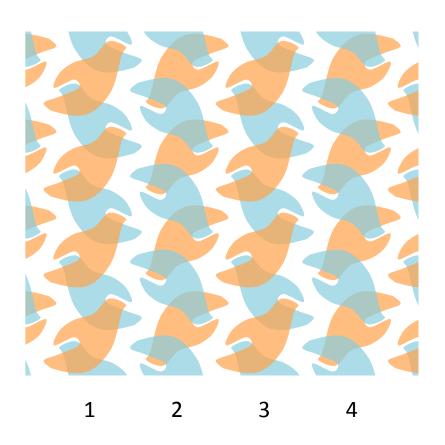












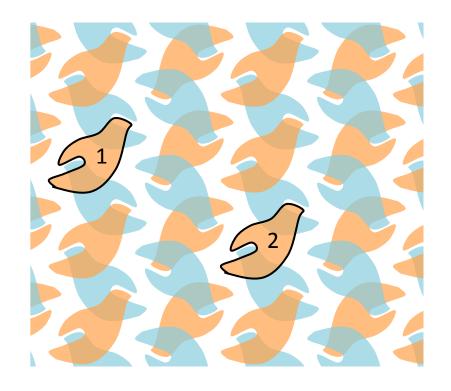
Orange and blue represent opposite sides of molecules

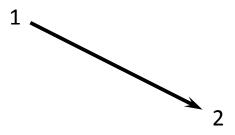
There is the third dimension.

A slice is shown, where

- column 1, 3 : orange-sided molecules on top
- column 2, 4: blue-sided molecules on top
- etc.

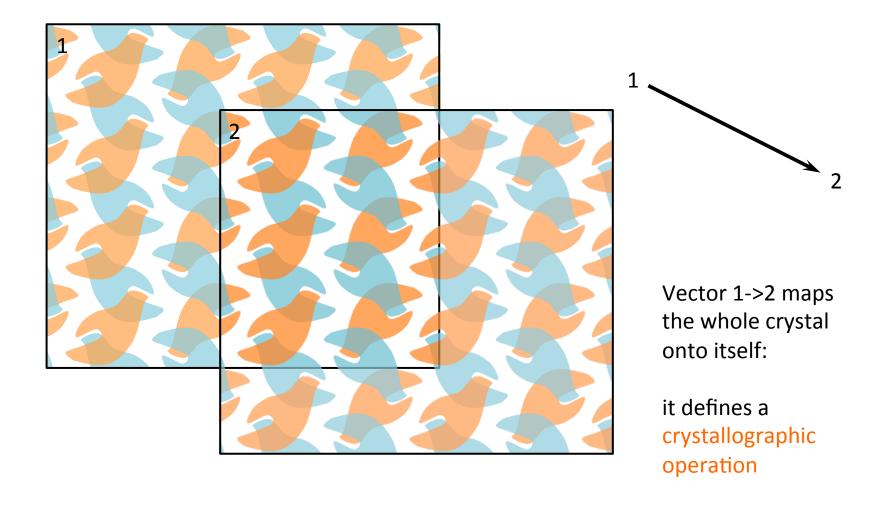
#### **Translation 1**



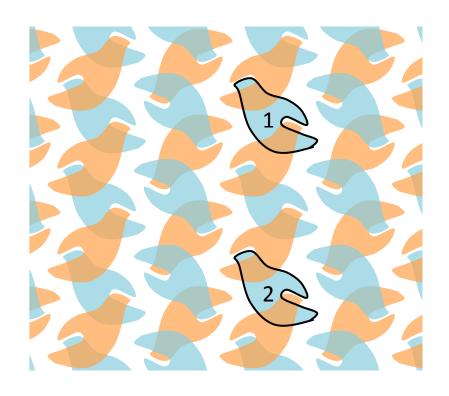


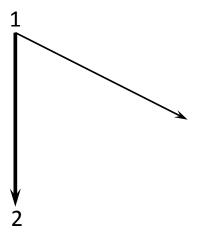
Vector maps 1 -> 2

### Translation 1 is global



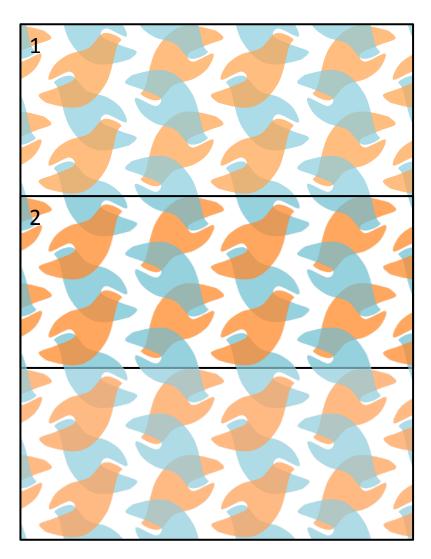
#### **Translation 2**

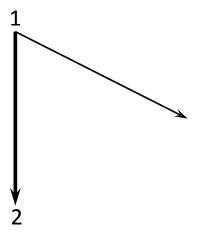




Highlighted vector maps 1 -> 2

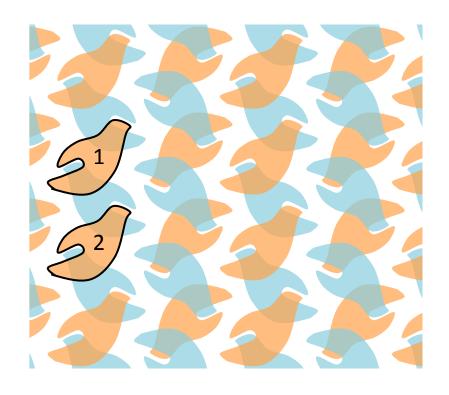
## Translation 2 is global

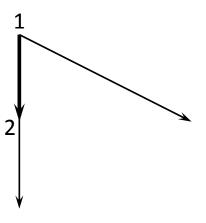




Highlighted vector maps the whole crystal onto itself

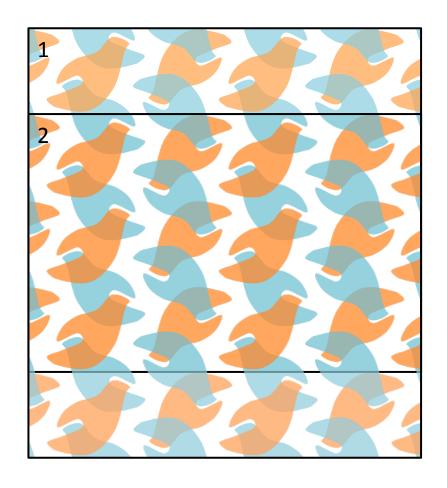
#### **Translation 3**

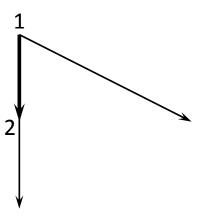




Highlighted vector maps 1 -> 2

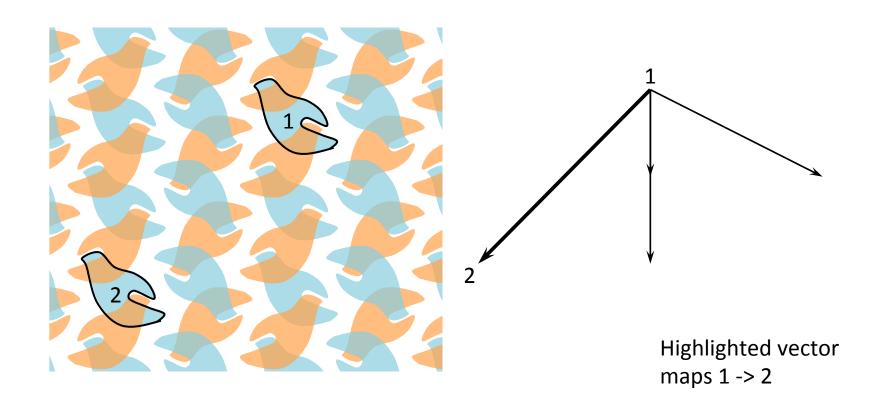
## Translation 3 is global



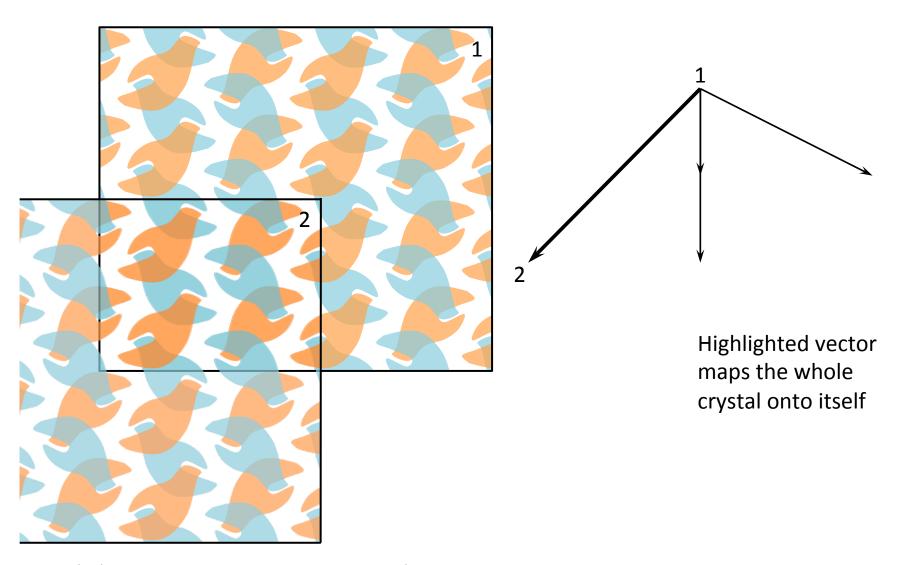


Highlighted vector maps the whole crystal onto itself

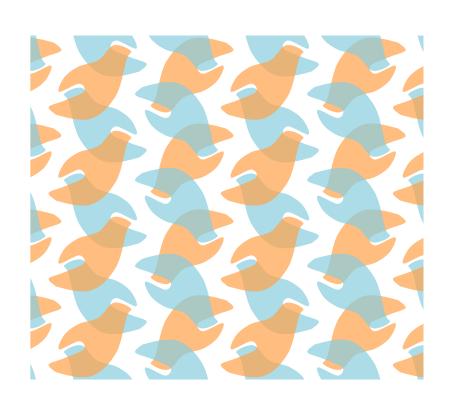
#### **Translation 4**

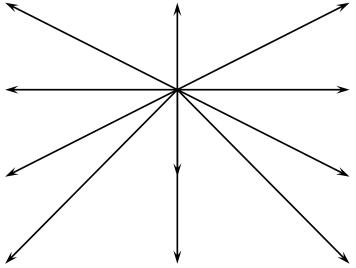


## Translation 4 is global



#### All translations form an infinite group

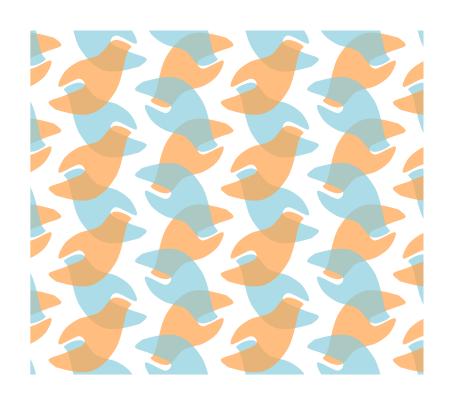


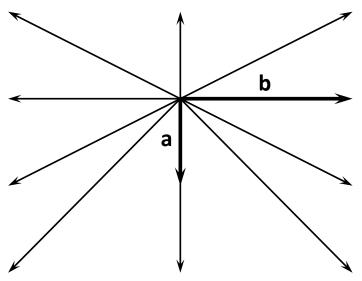


An infinite group (over sum):

- reverse translations included
- sum of any two vectors from the group belongs to the group

#### Basis set

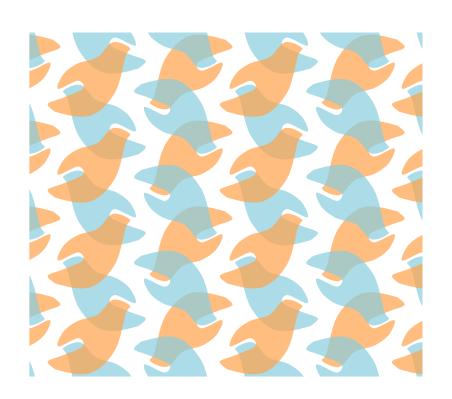


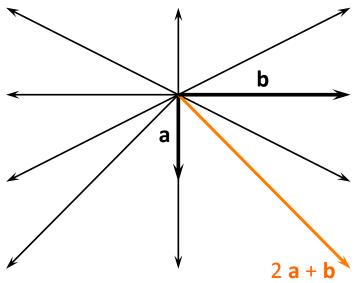


All the translations that map the crystal onto itself can be produced from a basis set: **a**, **b**, **c** 

(c is perpendicular to the plane)

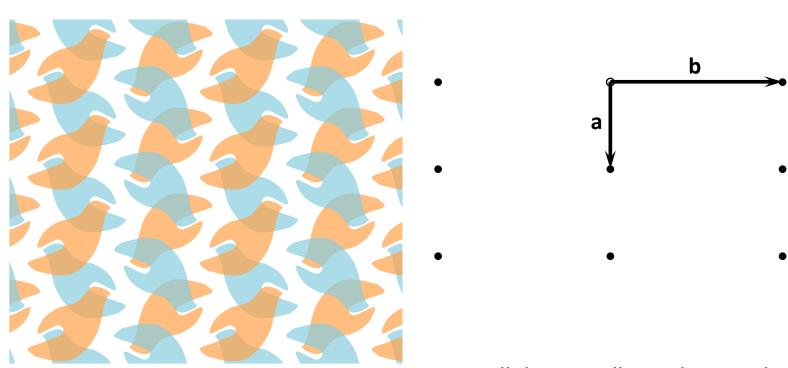
#### Basis set





For example, the highlighted vector is expressed as 2 **a** + **b**.

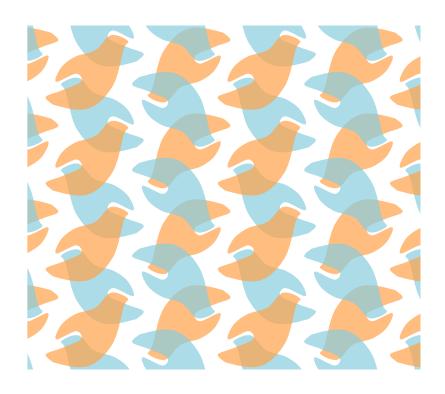
#### Lattice

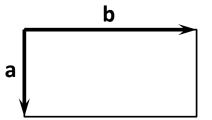


All the crystallographic translations can be represented as a lattice.

Translations live in a separate pace, not connected to crystal (for now)

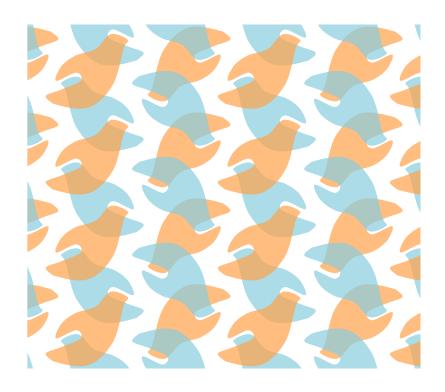
#### Unit cell

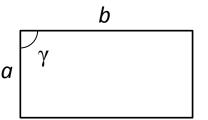




A compact representation of translational symmetry and base vectors.

#### Unit cell





Can be fully characterised by six numbers (the third dimension is not shown here)

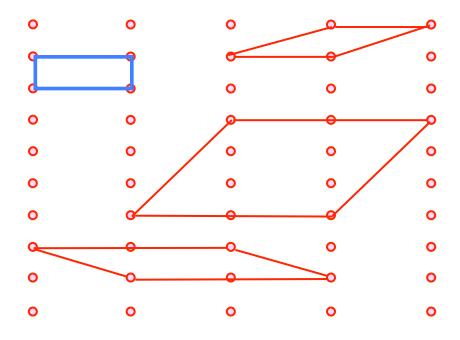
### Unit cell parameters (3D view)

Translation symmetry is defined Z by three base vectors **a**, **b**, and **c**. C a

Unit cells are usually defined in terms of the *lengths* of these vectors and angles between them. For example,

a=94.2Å, b=72.6Å, c=30.1Å,  $\alpha$ =90°,  $\beta$ =102.1°,  $\gamma$ =90°.

#### `Choice of unit cell

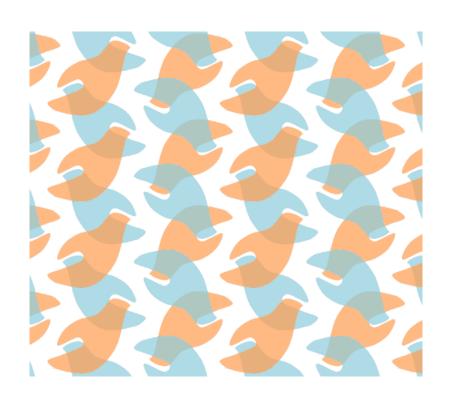


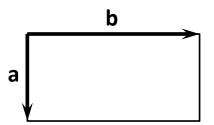
The top two do define all translations = primitive unit cells

The bottom two do NOT define all translations = non-primitive unit cells

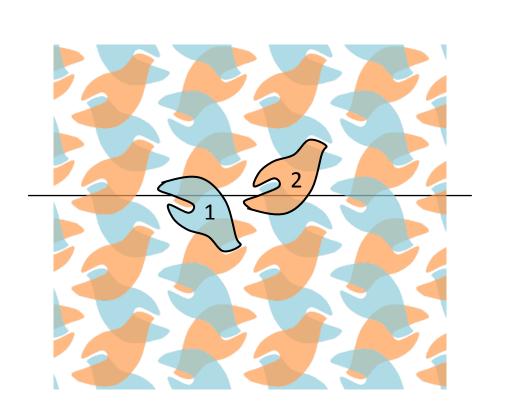
The top left: primitive reduced – the standard for <u>some</u> space groups (including  $P2_12_12_1$ )

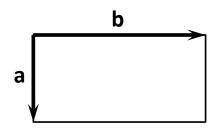
# Back to example

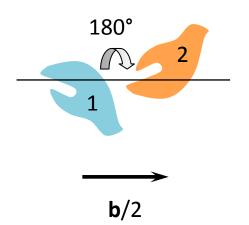




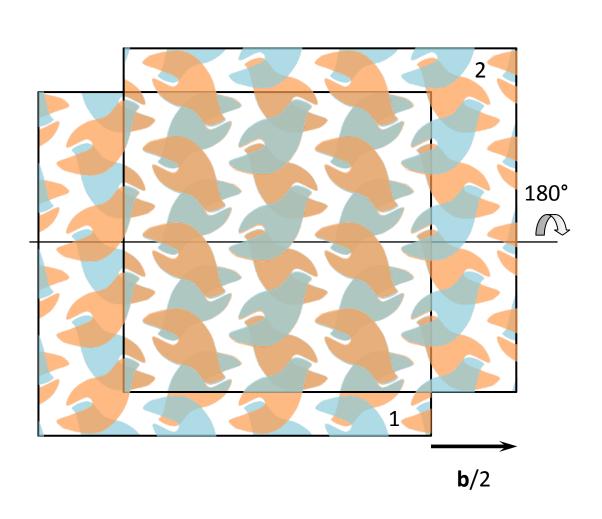
#### Screw rotation 1

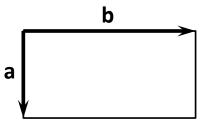






#### Screw rotation 1 is global





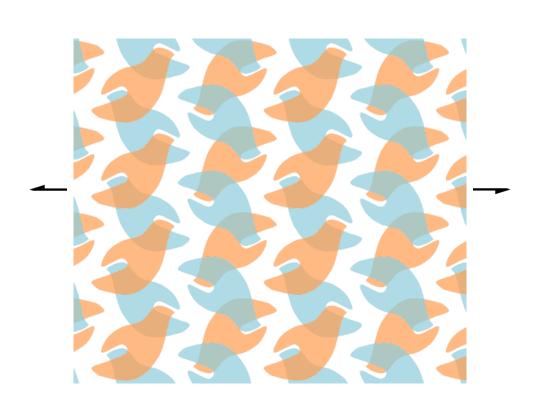
Operation 1->2 maps the whole crystal onto itself:

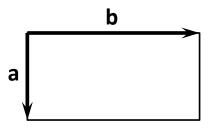
this is a crystallographic operation

The axis is a crystallographic symmetry element,

it can be mapped into the structure

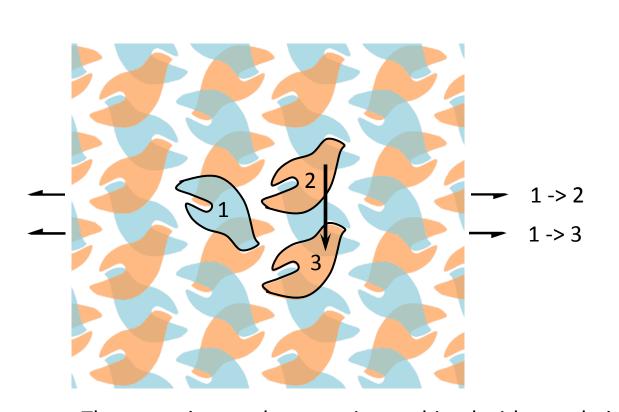
# Screw rotation 1 - symbol

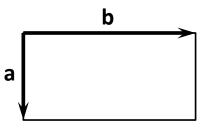




 $2_1$  (plane of figure): --

#### Screw rotation 1 - repeats





action of top axis

×

translation a

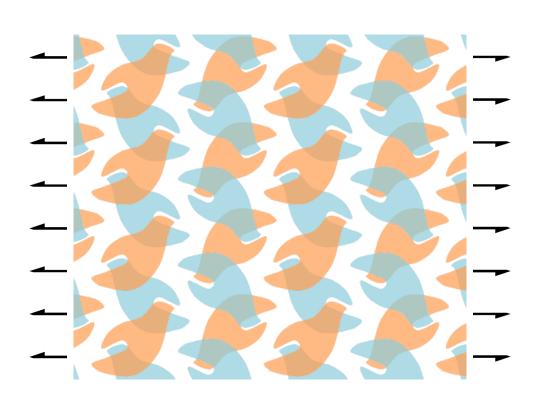
=

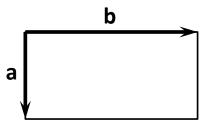
action of bottom axis

(elements of a group)

The operation on the top axis, combined with translation **a**, can be used to recreate the bottom axis. Here this also means that a rotation/translation offset by ½ **a** is also available.

#### Screw rotation 1 - repeats

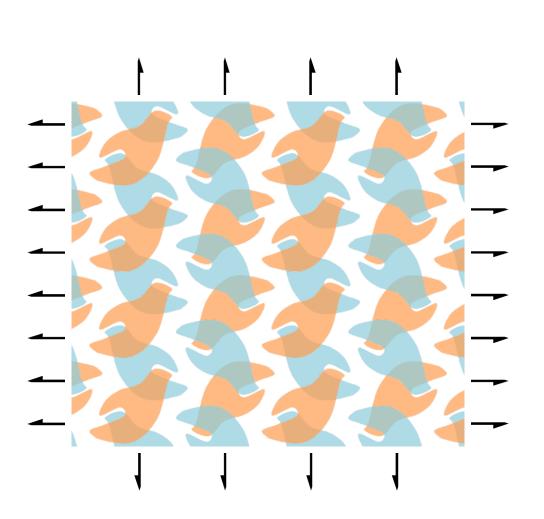


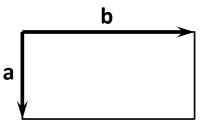


 $2_1$  (plane of figure): -

Also repeated in 3d dimension with offset of ½ **c** 

#### Screw rotations parallel to a and b

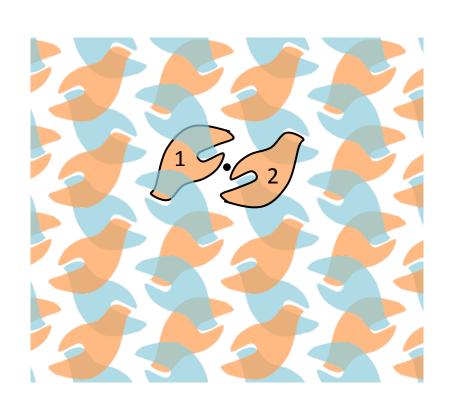


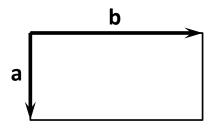


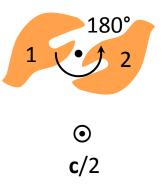
 $2_1$  (plane of figure): -

Series of 2<sub>1</sub> axes offset by ½ unit cell from each other.

#### Screw rotation 3 – into plane

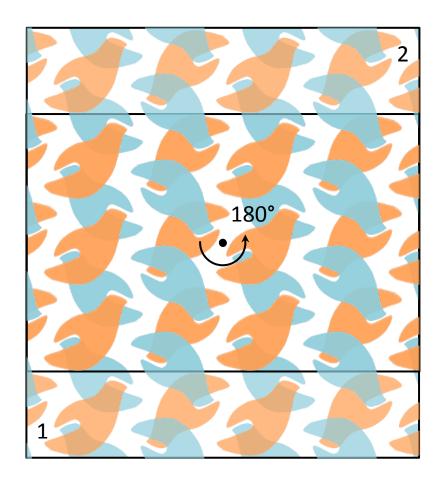




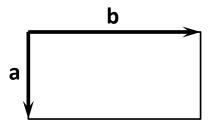


A rotation of 180° with a translation of ½ unit cell from the figure.

#### Screw rotation 3 is global







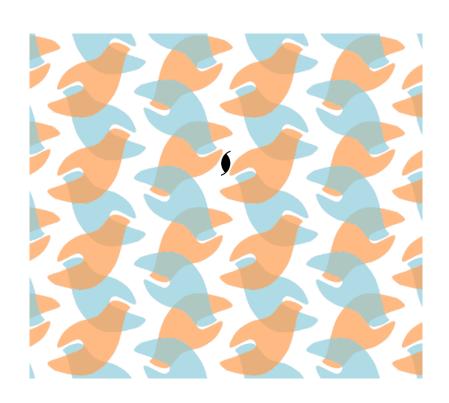
Screw rotation 3 maps the whole crystal onto itself:

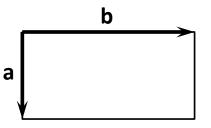
this is a crystallographic operation

The rotation axis is a crystallographic symmetry element,

it can be mapped into the structure

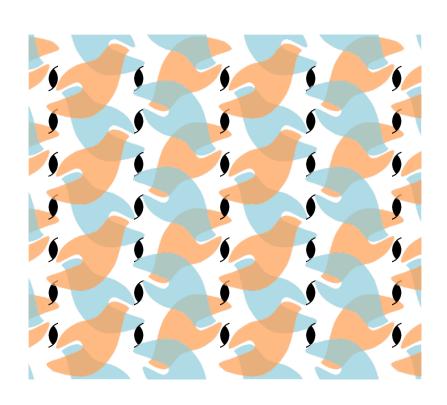
# Screw rotation 3 - symbol

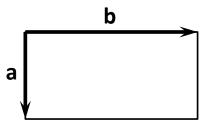




2<sub>1</sub> (along view):

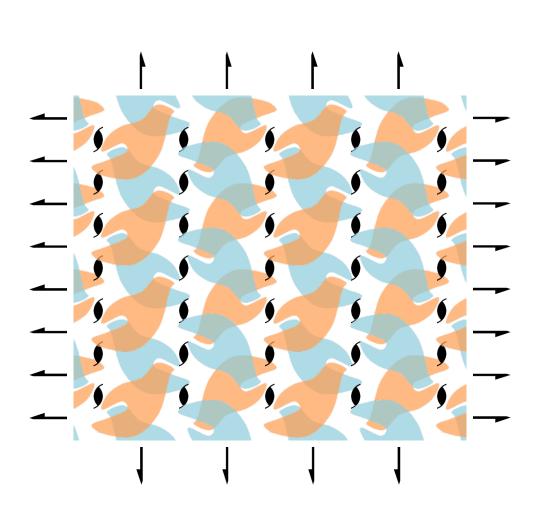
# Screw rotation 3 - repeats

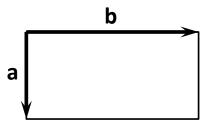




2<sub>1</sub> (along view):

### All axes together



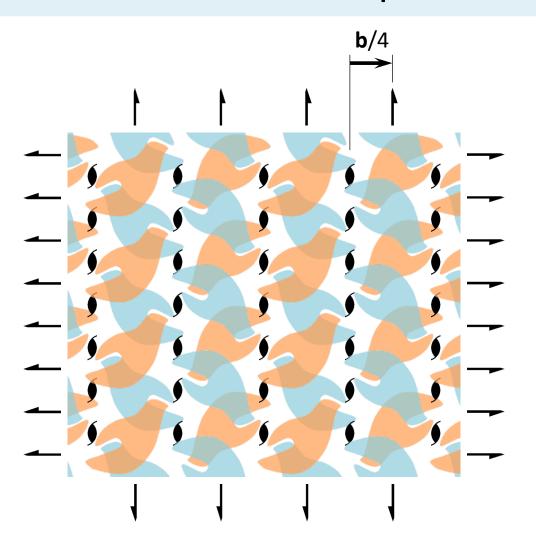


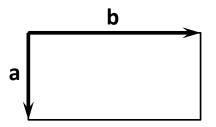
 $2_1$  (plane of figure): --

2<sub>1</sub> (along view):

we have built a space group

# Relative positions of axes

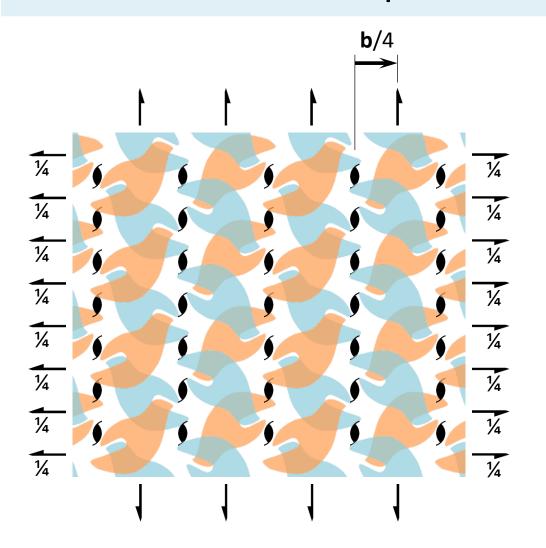


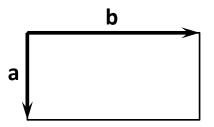


 $2_1$  (plane of figure): --

2<sub>1</sub> (along view):

#### Relative positions of axes





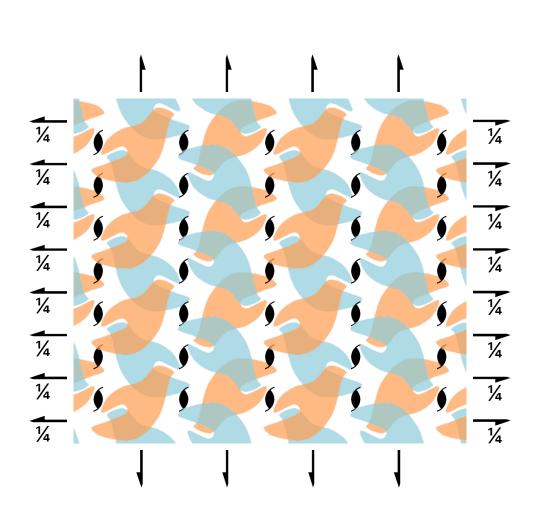
 $2_1$  (plane of figure): --

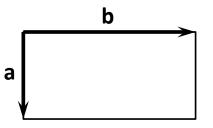
2<sub>1</sub> (along view):

The adjacent axes running in different directions are offset by ¼ unit cell edge from each other.

The horizontal ¼ indicates a offset of ¼ **c** into the figure.

# Relative positions of axes

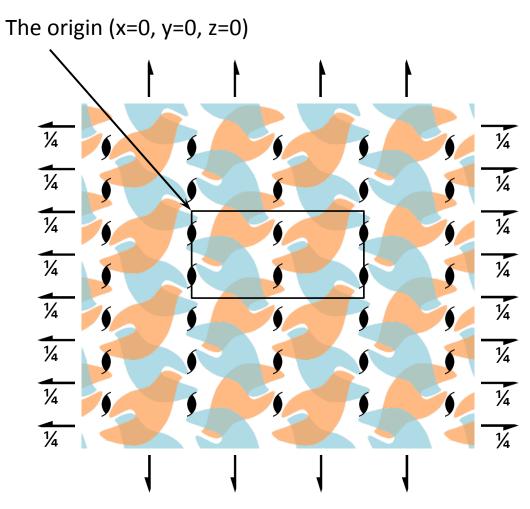




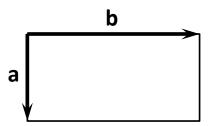
 $2_1$  (plane of figure): --

2<sub>1</sub> (along view):

### Choice of origin is a convention. Notation



The origin in this particular space group: is chosen to be equidistant from adjacent axes



 $2_1$  (plane of figure): --

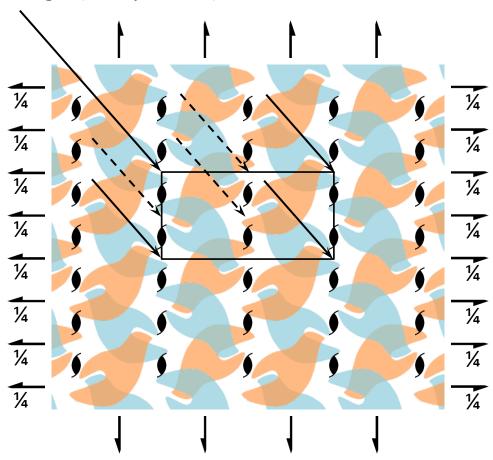
2<sub>1</sub> (along view):

The unit cell placed on picture with symmetry elements means a choice of origin.

Such a choice is a convention.

### Equivalent and alternative origins

The origin (x=0, y=0, z=0)



Solid arrows – origins, which are equivalent to the one chosen

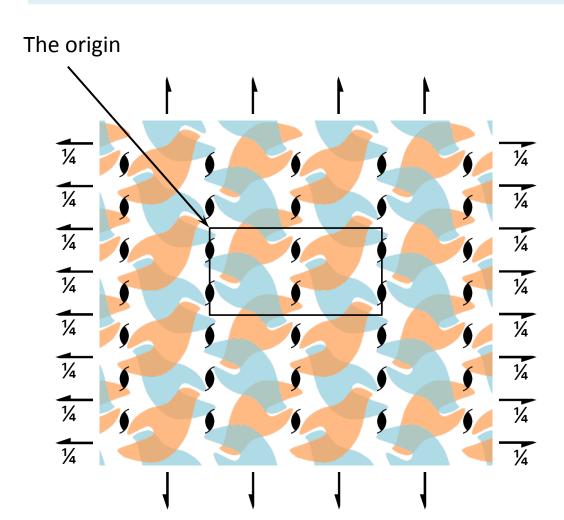
Dashed arrows – alternative origins.

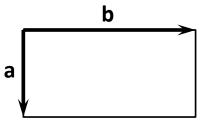
#### Altogether:

- infinite number of conventional origins
- eight types of equivalent origins in this example

The origin in this particular space group: is chosen to be equidistant from adjacent axes

# Complete picture



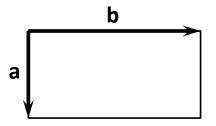


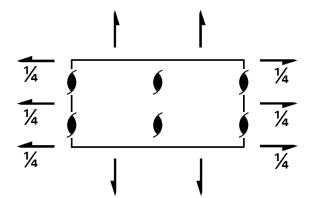
 $2_1$  (plane of figure): --

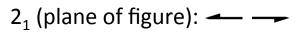
2<sub>1</sub> (along view):

#### Compact representation

$$P2_12_12_1$$
No. 19

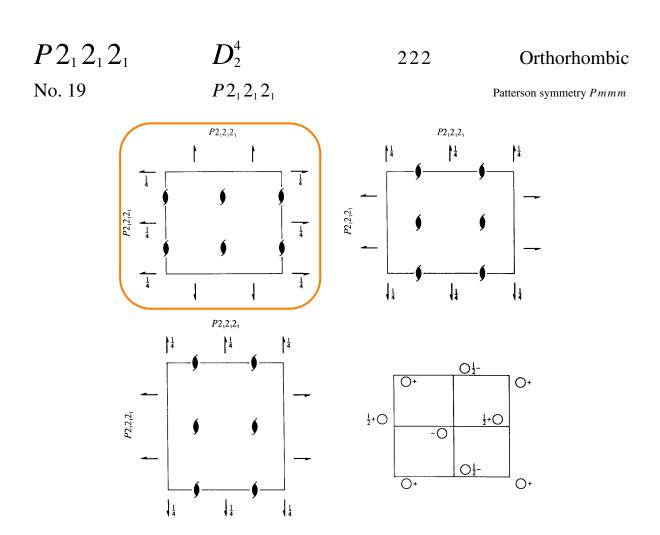




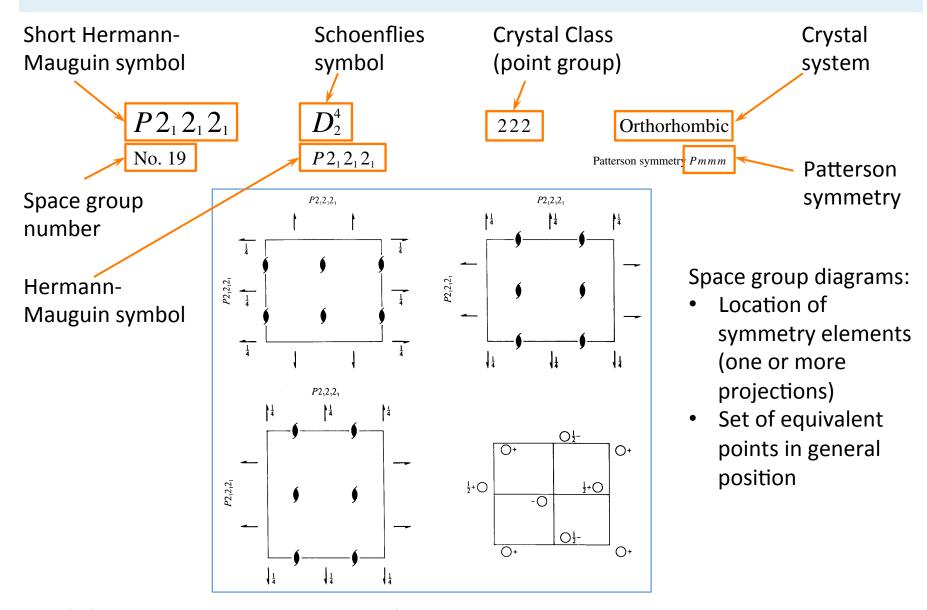


Compact representation -> space group symbol -> more info in International Tables

#### **Presentation in International Tables**



#### Presentation in International Tables



### Symmetry operations and elements

Apart from the identity and translational symmetry, macromolecular crystals can only contain the following symmetry elements:

Proper rotation: Rotate by 360°/n.

Screw rotation: Rotate by  $360^{\circ}/n$  & translate by d(m/n); d= unit cell edge.

Proper R	Rotations		Screw Rotations	
	Symbol	$(\mathbf{n})$	Symbol	$(\mathbf{n}_{m})$
Two-fold		2	<b>f</b>	$2_1$
Three-fold		3		$3_1, 3_2$
Four-fold		4		$4_1, 4_2, 4_3$
Six-fold		6		$6_1, 6_2, 6_3, 6_4, 6_5$

### Symmetry elements disallowed by chiral centres

Small molecules also face other symmetry operations

- Mirror plane m
- Glide planes **a**, **b**, **c**, **n** or **d**: reflection across plane followed by translation (usually ½) unit cell parallel to plane along **a**, **b**, **c**, **face diagonal** or **body diagonal**, respectively
- Rotation inversion  $\bar{1}, \bar{3}, \bar{4}, \bar{6}$ : a rotation followed by inversion

#### Space groups

- All possible combinations of symmetry elements => 230 space groups
- Because protein and nucleic acid molecules are chiral, there are only 65 "biological" space groups.
- Space groups are divided on 7 crystal system based on
  - the presence of symmetry elements of a certain order (6, 4, 3, 2)
  - the number of different orientations of these elements

### **Crystal Systems**

\* In macromolecular crystals the symmetry elements are all rotations

Crystal System	Characteristic symmetry elements	Convention
1. Triclinic	Translations only	
2. Monoclinic	2-fold axes, all parallel	along <b>b</b>
3. Orthorhombic	2-fold axes in three perpendicular directions	along a, b and c
4. Tetragonal	4-fold axes, all parallel	along <b>c</b>
5. Trigonal	3-fold axes, all parallel	along <b>c</b>
6. Hexagonal	6-fold axes, all parallel	along <b>c</b>
7. Cubic	3-fold axes in four different orientations	along body diagonals

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7. Cubic	3-fold axes in four different orientations	along body diagonals

example

# **Crystal Systems**

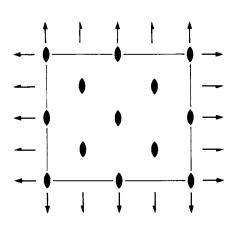
Some SGs require use of centred (non-primitive) unit cells —

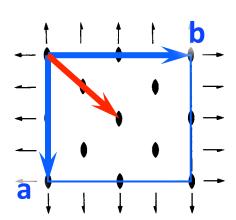
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2. Monoclinic	2-fold axes, all parallel	along <b>b</b>
3. Orthorhombic	2-fold axes in three perpendicular directions	along a, b and c
4. Tetragonal	4-fold axes, all parallel	along <b>c</b>
5. Trigonal	3-fold axes, all parallel	along <b>c</b>
6. Hexagonal	6-fold axes, all parallel	along <b>c</b>
7. Cubic	3-fold axes in four different orientations	along body diagonals

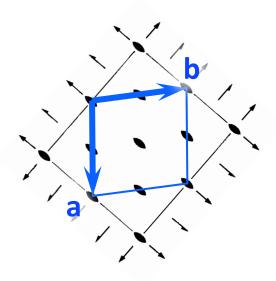
### C222: an example of a centred cell

C222
as presented in the
International Tables
for Crystallography

Standard unit cell; C means additional translation ½ (a + b) If we were using a primitive cell



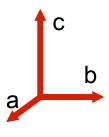




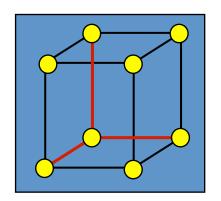
2-fold axes are along **a**, **b** and **c** (conventional setting)

Some 2-fold axes are along face diagonals (non-conventional crystal setting)

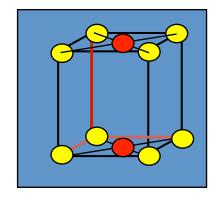
### Centred cells in pictures



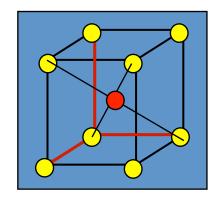
Also:
H - Hexagonal setting of rhombohedral space groups



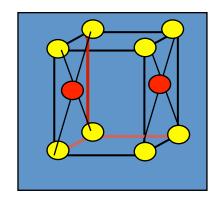
P – Primitive



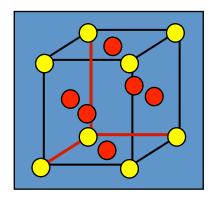
A – Face centred (A)



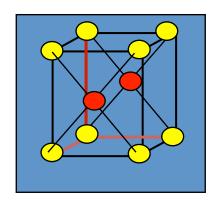
I – Body centred



B – Face centred (B)



F – Face centred (all)



C – Face centred (C)

#### **Bravais lattices**

- 7 crystal systems, combined with some of the centring types
   (P, C, I, F or H) gives 14 Bravais lattices
  - excluded are impossible combinations (e.g. A4)
  - or equivalent combinations (e.g. C4 and P4)

### **Bravais lattices**

Crystal System	Bravais Lattices
1. Triclinic	1. Primitive ( <i>P</i> )
2. Monoclinic	2. Primitive ( <i>P</i> ) 3. Base-Centered ( <i>C</i> )
3. Orthorhombic	<ul> <li>4. Primitive (P)</li> <li>5. Base-Centered (C)</li> <li>6. Body-Centered (I)</li> <li>7. Face-Centered (F)</li> </ul>
4. Tetragonal	8. Primitive ( <i>P</i> ) 9. Body-Centered ( <i>I</i> )
5. Trigonal	10. Primitive ( <i>P</i> )
	11. Rhombohedral ( <i>R or H</i> )
6. Hexagonal	10. Primitive ( <i>P</i> )
7. Cubic	12. Primitive ( <i>P</i> ) 13. Body-Centered ( <i>I</i> ) 14. Face-Centered ( <i>F</i> )

example

#### Triclinic P 1

"P" means primitive lattice type

"1" means no symmetry operations except for translations

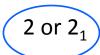
No constraints on a, b, c,  $\alpha$ ,  $\beta$ ,  $\gamma$ 

#### Monoclinic

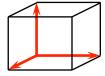
$$P 1 2 1 P 1 2_1 1$$
  
 $C 1 2 1$ 

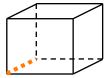
P or C

1



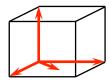
1



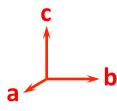








"1" means no symmetry axes in a given direction "2" or "2<sub>1</sub>" means 2-fold axes in a given direction



$$\alpha = \gamma = 90^{\circ}$$

Note: by convention the 2-fold is along **b** (other settings are sometimes used as well)

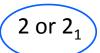
#### Orthorhombic

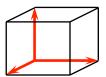
$$P \ 2 \ 2 \ 2$$
  $P \ 2 \ 2 \ 2_1$   $P \ 2_1 \ 2_1 \ 2$   $P \ 2_1 \ 2_1 \ 2$   $P \ 2_1 \ 2_1 \ 2$   $P \ 2_2 \ 2$   $P \ 2_2 \ 2$   $P \ 2_1 \ 2_1 \ 2$ 

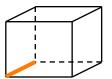
P, C, I or F



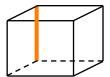


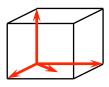




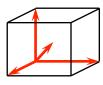


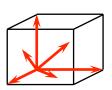


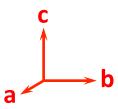




"2" or "2<sub>1</sub>" means 2-fold axes in a given direction







$$\alpha = \beta = \gamma = 90^{\circ}$$

#### **Tetragonal**

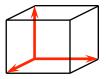
$$P 4 2_1 2 P 4_1 2_1 2 P 4_2 2_1 2 P 4_3 2_1 2  $P 4 2 2 P 4_1 2 2 P 4_2 2 2 P 4_3 2 2  $I 4 2 2 I 4_1 2 2$$$$

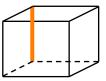
P or I

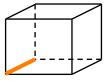


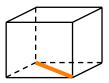
2, 2<sub>1</sub> or None

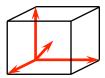
2 or None

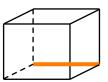


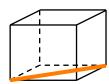


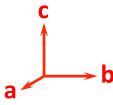












$$\alpha = \beta = \gamma = 90^{\circ}$$
 $a = b$ 

 $a \equiv b$  due to the 4-fold relating them

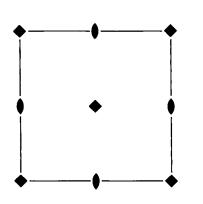
All 2-fold axes are also related via 4-fold rotations and either

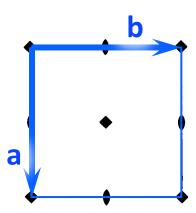
- all of them are present or
- none of them are present

#### C4: an example of a redundant space group symbol

*P*4 as presented in the **International Tables** for Crystallography

and with base vectors shown

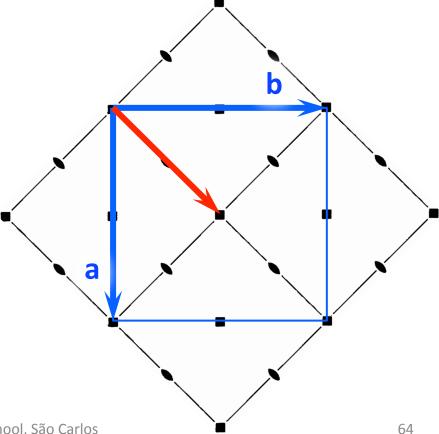




*C*4

This is a valid, but redundant spacegroup as it is obtained from P4

- by rotation 45° and
- redefining base vectors
- additional translation (a + b)/ 2



### **Trigonal**

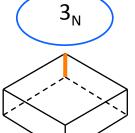
$$P \ 3 \ 2 \ 1$$
  $P \ 3_1 \ 2 \ 1$   $P \ 3_2 \ 2 \ 1$   $P \ 3_1 \ 2$   $P \ 3_2 \ 1 \ 2$ 

P3 P3<sub>1</sub> P3<sub>2</sub>

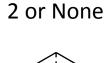
H 3 2

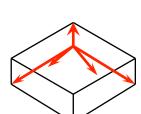
*H*3

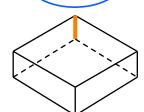
*P* or *H*(\*)

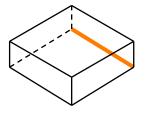


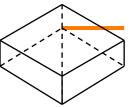
2 or None

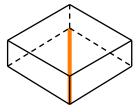


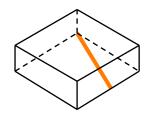


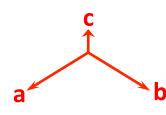




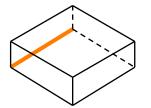


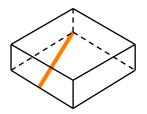






$$\alpha = \beta = 90^{\circ}$$
 $\gamma = 120^{\circ}$ 
 $a = b$ 





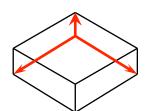
(\*) an alternative rhombohedral (R) is also used

#### Hexagonal

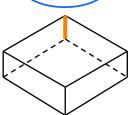
P 6 2 2  $P 6_1 2 2$   $P 6_2 2 2$   $P 6_3 2 2$   $P 6_5 2 2$   $P 6_4 2 2$ 

 $P6 P6_1 P6_2 P6_3$   $P6_5 P6_4$ 

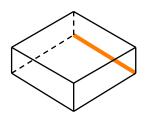
Р

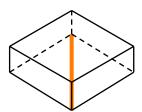


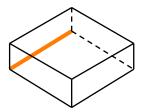
 $6_N$ 



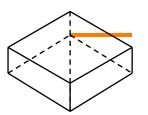
2 or None

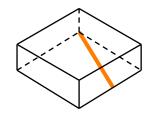


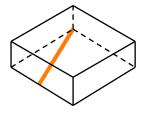


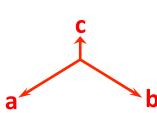


2 or None









 $\alpha = \beta = 90^{\circ}$   $\gamma = 120^{\circ}$  a = b

#### Cubic

P432  $P4_132$   $P4_232$   $P4_332$  I432  $I4_132$  F432  $F4_132$ 

P, I or F

 $4_N$  or  $2_N$ 

3

2 or None





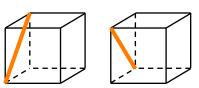








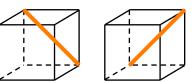


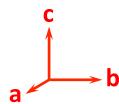












$$\alpha = \beta = \gamma = 90^{\circ}$$
 $a = b = c$ 



# Monoclinic (lattice based setting)

$$\alpha = \gamma = 90^{\circ}$$

additional condition:  $\beta < 120^{\circ}$ 

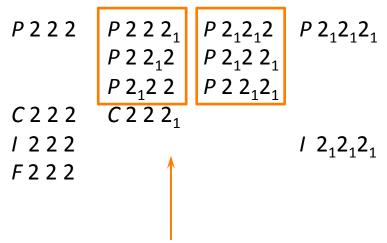
In orange frame:

- the same space group
- different crystal setting

# Orthorhombic (lattice based setting)

$$\alpha = \beta = \gamma = 90^{\circ}$$

additional condition: a < b < c



In orange frames:

- the same space group
- different crystal setting

### Rules in a Table

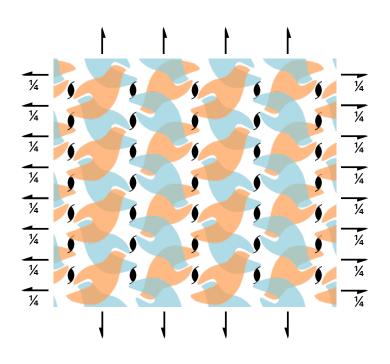
Crystal System	Characteristic symmetry elements	Bravais Lattices	Unit Cell Geometry
1. Triclinic	None	1. Primitive ( <i>P</i> )	$a \neq b \neq c;$ $\alpha \neq \beta \neq \gamma$
2. Monoclinic	2-fold axes, all parallel	2. Primitive ( <i>P</i> ) 3. Base-Centered ( <i>C</i> )	$a \neq b \neq c;$ $\alpha = \gamma = 90^{\circ} \neq \beta$
3. Orthorhombic	2-fold axes in three perpendicular directions	4. Primitive ( <i>P</i> ) 5. Base-Centered ( <i>C</i> ) 6. Body-Centered ( <i>I</i> ) 7. Face-Centered ( <i>F</i> )	$a \neq b \neq c;$ $\alpha = \beta = \gamma = 90^{\circ}$
4. Tetragonal	4-fold axes, all parallel	8. Primitive ( <i>P</i> ) 9. Body-Centered ( <i>I</i> )	$a = b \neq c;$ $\alpha = \beta = \gamma = 90^{\circ}$
5. Trigonal	3-fold axes, all parallel	10. Primitive ( <i>P</i> ) 11. Rhombohedral ( <i>H</i> - setting)	$a = b \neq c;$ $\alpha = \beta = 90^{\circ}, \ \gamma = 120^{\circ}$
6. Hexagonal	6-fold axes, all parallel	10. Primitive ( <i>P</i> )	$a = b \neq c;$ $\alpha = \beta = 90^{\circ}, \ \gamma = 120^{\circ}$
7. Cubic	3-fold axes in four different orientations	12. Primitive ( <i>P</i> ) 13. Body-Centered ( <i>I</i> ) 14. Face-Centered ( <i>F</i> )	a = b = c; $\alpha = \beta = \gamma = 90^{\circ}$

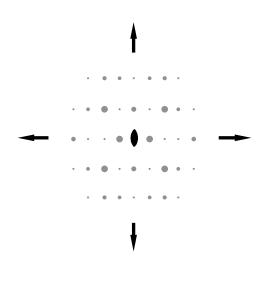
### Symmetry of intensities

The concept of reciprocal lattice is based on angular relation between the incident beam and the Bragg planes. Therefore:

- Reciprocal lattice rotates together with crystal
- However, reciprocal lattice is not translated together with crystal

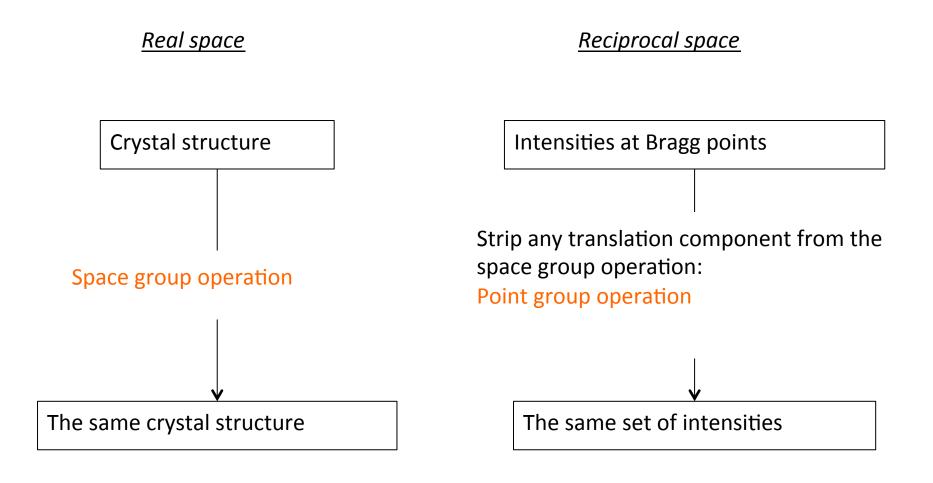
### Symmetry of intensities





All axes of the same order and in the same direction are "merged" together to give an element of a point group.

### Symmetry of intensities



## Space group and point group

#### Crystal space group

Arithmetic crystal class

International Tables for Crystallography (2006). Vol. A,

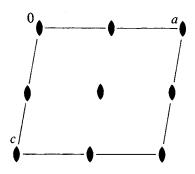
*P*2

 $C_2^1$ 

No. 3

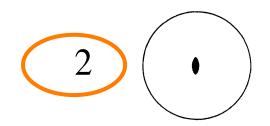
P121

UNIQUE AXIS b



2*P* 

Crystal point group



# Space group and point group

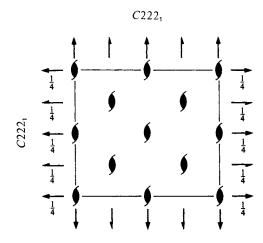
#### Crystal space group

#### Arithmetic crystal class

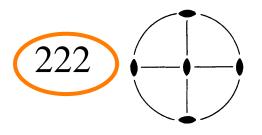
International Tables for Crystallography (2006). Vol. A,



222*C* 



Crystal point group

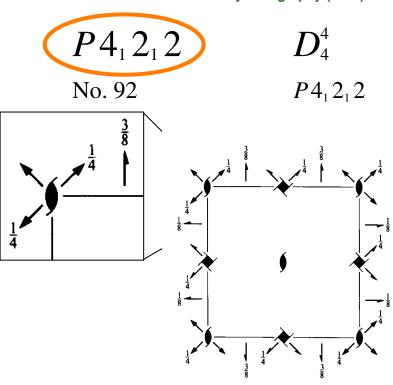


# Space group and point group

#### Crystal space group

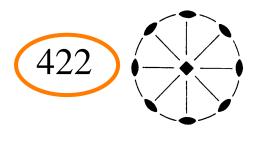
#### Arithmetic crystal class

International Tables for Crystallography (2006). Vol. A,

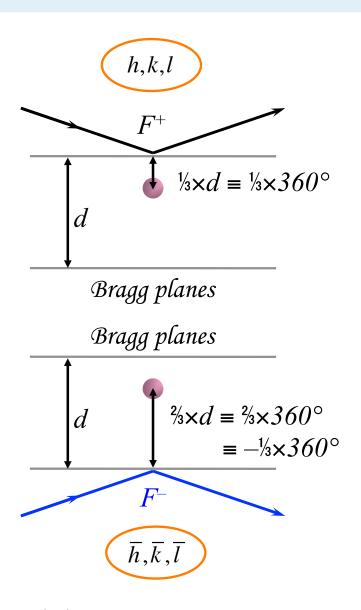


422*P* 

Crystal point group



#### Friedel's law



- Bragg planes
  - define reference phase
- Probe atom:

$$\Delta\varphi\left(\overline{h},\overline{k},\overline{l}\right) = -\Delta\varphi\left(h,k,l\right)$$

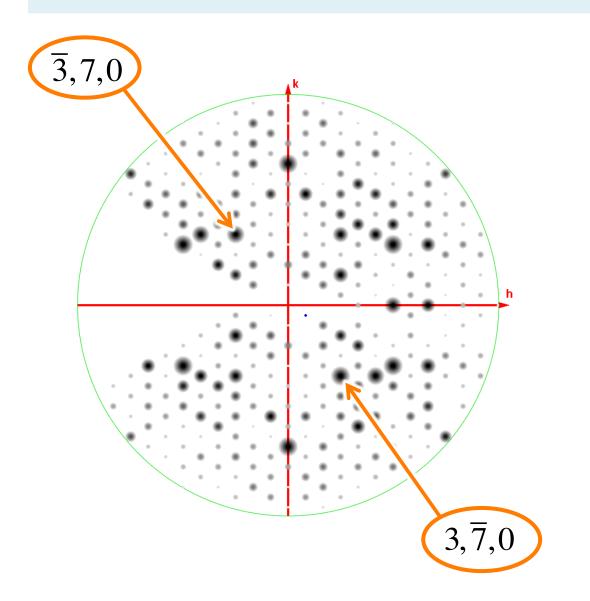
$$\Delta F(\overline{h}, \overline{k}, \overline{l}) = \Delta F^*(h, k, l)$$

• Total:

$$F(\overline{h},\overline{k},\overline{l}) = F^*(h,k,l)$$

$$I(\overline{h},\overline{k},\overline{l}) = I(h,k,l)$$

### Friedel's law



$$I(\overline{3},7,0) = I(3,\overline{7},0)$$

## Point group and Laue group

+ inversion =

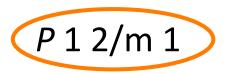
Arithmetic crystal class

2*P* 

Crystal point group

2

Patterson space group



Laue point group

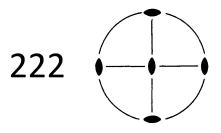
## Point group and Laue group

+ inversion =

Arithmetic crystal class

222*C* 

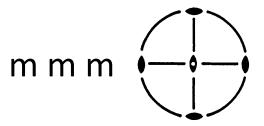
Crystal point group



Patterson space group



Laue point group



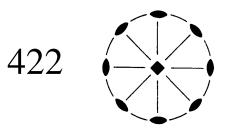
## Point group and Laue group

+ inversion =

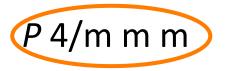
Arithmetic crystal class

422*P* 

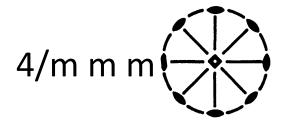
Crystal point group



Patterson space group



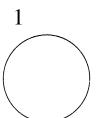
Laue point group



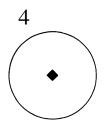
## The eleven point groups or crystal classes

Crystal system	Laue point group	Non-centrosymmetric point groups belonging to the Laue point group
Cubic	m3m m3	432 43 <i>m</i> 23
Tetragonal	4/mmm 4/m	$\begin{array}{ccccc} 422 & 4mm & \overline{4}2m \\ 4 & \overline{4} \end{array}$
Orthorhombic	mmm	222 mm2
Trigonal	3 <i>m</i> 3	32 3 <i>m</i> 3
Hexagonal	6/mmm 6/m	$\begin{array}{cccc} 622 & 6mm & \overline{6}m2 \\ 6 & \overline{6} \end{array}$
Monoclinic	2/m	2 m
Triclinic	1	1

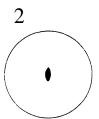
# The point groups that can exist in protein crystals

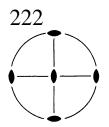


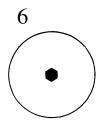
If it helps view as sphere

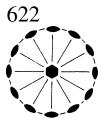


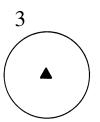
422

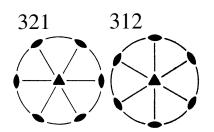


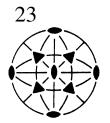


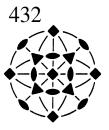




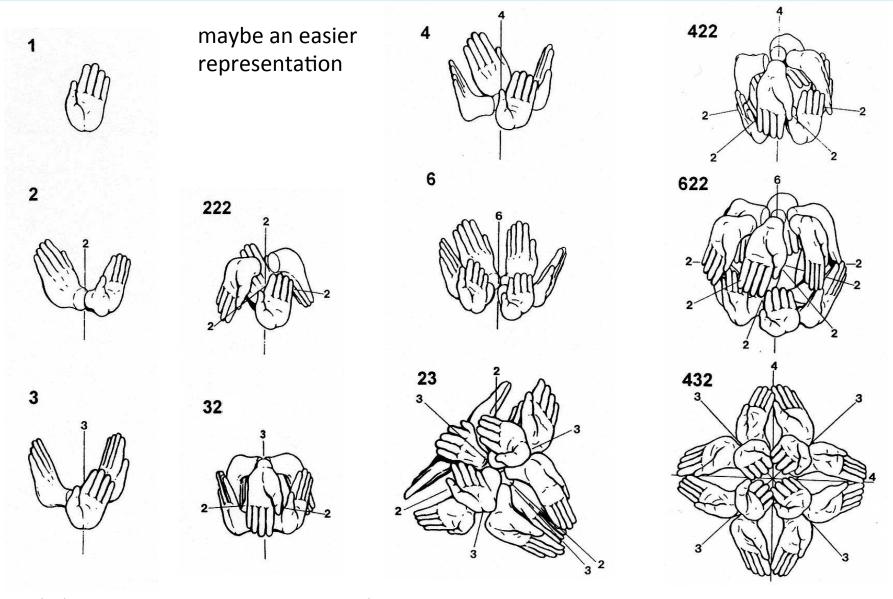








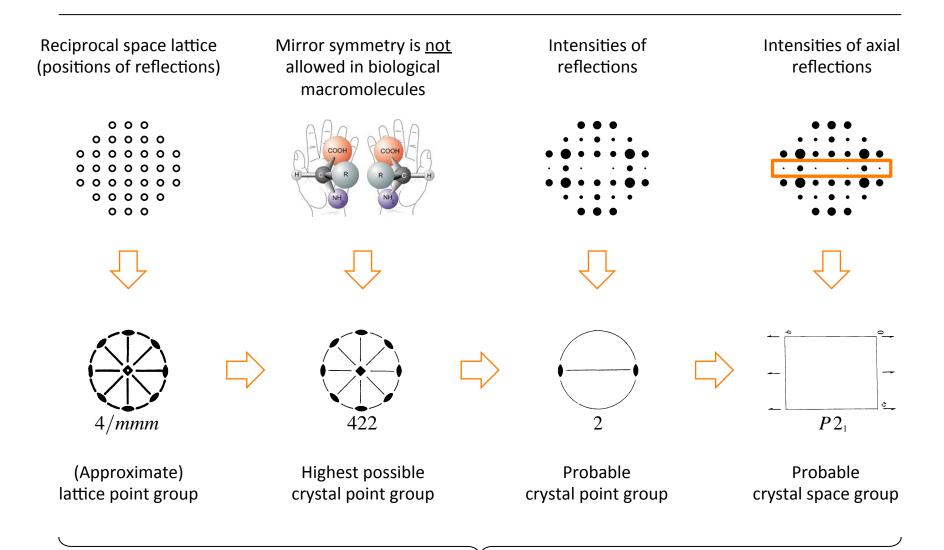
# The point groups that can exist in protein crystals



### How do we deduce the Space Group in practice?

- We start in reciprocal space (point group)
- We go all way back from symmetry in reciprocal space to crystal space group
  - Data processing gives values of the unit cell parameters
  - Lattice symmetry is derived from the unit cell parameters
  - Comparison of related intensities gives crystal point group
  - Systematic absences allow to reduce the number of possible space groups.
  - Space group is only a hypothesis until structure is complete

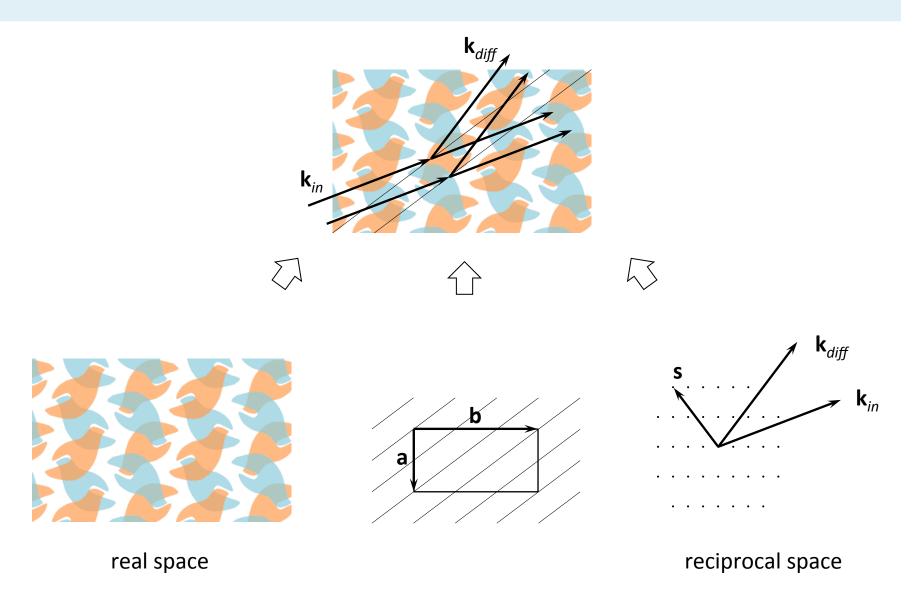
#### Space group assignment (e.g. Pointless)



User: decision making, structure solution, final space group assignment

# End

### Conventional diffraction scheme



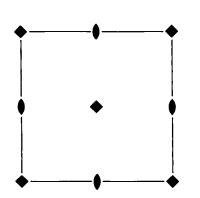
#### C4: an example of a redundant space group

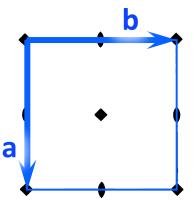
P4
as presented in the
International Tables
for Crystallography

Standard unit cell; C means additional translation **a** + **b**  *C*4

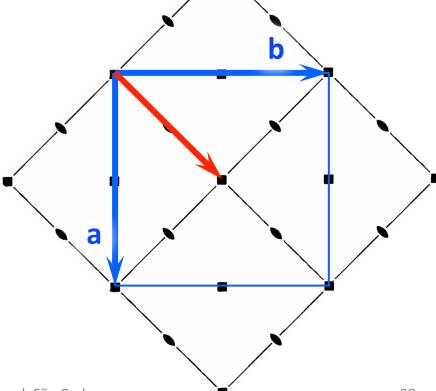
This is a valid, but redundant spacegroup as it is obtained from P4 by

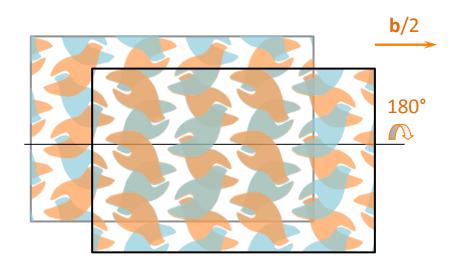
rotation 45° and redefining base vectors

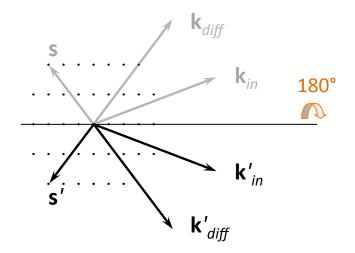




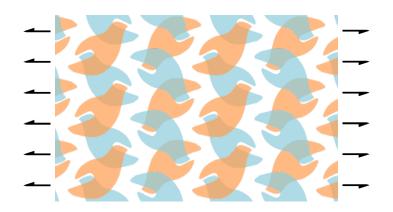
A new slide; it needs a bit of thinking and some reforamtting + one more slide on impossible combination of crystal class and centring type

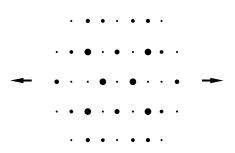






$$I(s') = I(s)$$





#### **Monoclinic**

(lattice based setting)

$$\alpha = \gamma = 90^{\circ}$$

additional condition:  $\beta < 120^{\circ}$ 

In orange frame:

- the same space group
- different crystal setting

### Orthorhombic

(lattice based setting)

$$\alpha = \beta = \gamma = 90^{\circ}$$

additional condition: a < b < c





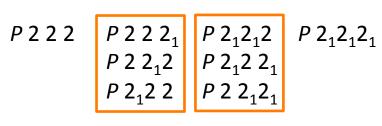
In orange frames:

- the same space group
- different crystal setting

# Orthorhombic (lattice based setting)

$$\alpha = \beta = \gamma = 90^{\circ}$$

additional condition: a < b < c





In orange frames:

- the same space group
- different crystal setting

## Centred (non-primitive) unit cells

#### Highlighted in orange

- Triclinic: P1, P1
- *Monoclinic*: P2, P2<sub>1</sub>, C2, ...
- Orthorhombic: P222, P222<sub>1</sub>, P2<sub>1</sub>2<sub>1</sub>2, P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, C222, C222<sub>1</sub>, F222, I222, I2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, ...

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